

Foran to Foran.

TO F. J. FORAN, MERCHANT TAILOR, CHARLOTTE-TOWN, FROM J. K. FORAN, OTTAWA.

Art thou poetic, too, my friend?

Because we bear the one good name, Must both towards high Parnassus tend? Must both our spirits be the same?

'Tis passing strange that we should both Attempt to woo the Graces Nine, And neither you nor I are loath To touch the string of round Divine.

But still, methinks you do invade The region where I'm bound to reign; I've never envied your good trade, Why then compete for my domain?

Even here beside the lordly wave, That plashes 'neath our stately hill, Where Indian heroes—true and brave— Displayed their native pride and skill.

I've heard of your far-reaching fame, Your matchless stock, your goody store, And proud was I to see your name Above all competition soar.

Permit me then to wish you well, Success in life, the joys of love! Such is the wish my verse would tell— Success below and joy above!

Although perchance we'er may meet, This side of you dread mystic land; Yet souls can her, each other greet, In spirit, then—I shake your hand.

JOSEPH K. FORAN.

House of Commons, Ottawa, 18th April, 1889.

Gypsum and its Uses.

AN INTERESTING PAPER BY DAVID BROWN OF PARIS, ONT.

The value of gypsum or plaster of Paris, on certain soils, is well known to many farmers. In wet seasons the benefit is immediately apparent in increased yields of red clover, roots and other crops; but even when the results are not very noticeable at first, as in dry seasons, careful observation and comparison of fields sown with the plaster, and fields not so sown, show that the gypsum often repays many hundred per cent. the cost of buying and sowing it. Mr. Dryden, M. P. P., found a single dressing of Paris plaster operating distinctly for seven years on a field on his farm in Whitchy township, and similar testimony of equally careful observers shows like results elsewhere. Considering the importance of the fertilizer and its frequent use in Ontario, we give this week part of the paper on Gypsum, read by David Brown, of Paris, a gentleman thoroughly acquainted with the subject, at the recent Farmers' Institute at Paris, Ont. Mr. Brown said:—

"It is in accordance with the fitness of things that the subject of gypsum, or sulphate of lime, should be treated upon, at a meeting of the North Brant Farmers' Institute, held in the town of Paris. The name 'Paris' is understood to have been adopted because of the beds of Paris plaster which are to be found in this locality, the suggestion having come from the similarity of formation which underlies that great city upon the banks of the Seine—Paris, the capital of France. Beds of gypsum are only known to exist in Ontario, from the vicinity of Cayuga running along the valley of the Grand River to this point. The beds are found at various depths below the surface of the earth, at Cayuga, Mount Healey and Caledonia the depth is from 50 to 60 feet, and at Paris one mine is almost 90 feet below the highest elevation of ground, and in others it is considerably less. All of the working beds in the locality of Paris are approached by drifts which are dug from a level about seven or eight feet above the bed of the Grand River. Mines in this section have been worked for about sixty years, and the best authorities claim that the length of underground tunnelling would reach, if placed in a line, a distance of over nine miles. Many of these drifts or tunnels have been closed during the course of years, the superincumbent mass of rock and earth above having crushed the timber supports which were used to keep the drift in proper shape. Boring for salt and petroleum in the counties of Lambton and Huron, showed beds of gypsum at a depth of several hundred feet in that region, but no practical use could be made of these deposits for their inaccessible nature.

"In the early part of the present century the settlers drove long distances from all parts of this western country to obtain the mineral, then mined at what was called the 'Forks of the Grand River,' at the point where Smith's creek enters the river. The masses of rock gypsum were removed by sledge hammers as finely as possible by a process, and the powder thus obtained was scattered upon their fields by these old pioneers, who were quick to recognize the great virtue and usefulness of this fertilizer. The virgin soil, though rich in the other constituents of plant food, seemed to require the additional stimulus of this sulphate to promote vegetation. Gypsum, scientifically known as sulphate of lime, has for its constituent proportion: Lime, 32.5%; sulphuric acid, 46.51, and water, 20.93. The colors of gypsum are in great variety. Dark, white, yellowish white, grey, red, brown or black, according to its purity. The discoloration, when caused by the filtration of water through soil above, very frequently partakes of the properties of the soil, and when pure white it is the more highly crystallized. Its solubility depends very largely upon the state of crystallization. Six-sided and three-sided prisms, etc., and also the fibrous variety known to miners under the name of 'comb' gypsum, it is wholly useless for the purpose of stucco making and land plaster, although prized by the geologist or scientist for the peculiarities of the formation. White gypsum ground and boiled until the water of crystallization is evaporated, is the stucco, or plaster of Paris used by the artist, and the plaster for decorating and furnishing the walls of buildings, and nothing in the arts has yet been discovered to supersede its use for these purposes. In another form, ground to an impalpable powder, heated with chemicals, and colored to any shade which fancy may demand, it assumes the forms of alabaster, and makes the best, most healthful and permanent wall finish now in use in Canada.

It is, however, chiefly of the use of gypsum, in the form of land plaster, or as a fertilizer, that we are called upon to treat on this paper. The rock undergoes the simple process of grinding by means of burr stones, and it is proper here to remark that upon the fitness of the article thus produced its successful use largely depends. To its solubility in water alone must be ascribed its value as a manure, and when it is mentioned that in order to its perfect dissolving, 461 parts of water to one of gypsum are required, the necessity of perfect grinding must be at once apparent. Chemists have advanced a great variety of opinions regarding the action of gypsum upon the crops, and I quote a few of them for the information of the association. 'According to Koller the action of gypsum depends upon the power possessed by lime to form with the oxygen and carbon of the atmosphere, compounds which are fertile in vegetation.

Ruckert thinks it acts like any other food, Mayer and Braun say it merely improves the physical properties of the soil. Hedwig calls gypsum the saliva and gastric juice of plants. Humphrey Davy regarded it as an essential constituent of plants, because it acts only where gypsum is wanting in the soil. Liebig says it fixes the ammonia of the atmosphere. Johnson says that, if the land be deficient in lime, the gypsum will act, not only by virtue of the sulphuric acid, but of the lime which it contains. Bell says it 'acts in two ways, first, by applying sulphuric acid or sulphur and lime in a soluble state. Second, by absorbing ammonia and making a double compound with some alkaline salts.

Upon what kind of soil will gypsum show the best results? The answer must come: 'Upon land which is deficient in lime. Many soils have an abundance of lime, and upon these gypsum shows the minimum result, because the lime which it contains is not required. Neither is the moisture which the sulphuric acid attracts required when the plant is making its strongest growth. The conclusion then is that soils which are deficient in lime will show the best effects from the use of plaster. It is well always to remember that land plaster only furnishes two constituents of plant food, viz., sulphur and lime. Its action in attracting moisture can be comprehended in some measure when we think of the rapidity with which sulphuric acid left standing in an unworked bottle will absorb moisture. You will notice that after a few days the quantity in the bottle will have increased. Along with this moisture the acid has absorbed ammonia, which permeates the atmosphere, and thus indirectly the plant is furnished with that most powerful of stimulants which promotes its growth, the water working conjointly with the ammonia, conveying it to the roots and producing that freshness and greenness which is so strikingly an evidence of the healthful condition of a crop.

"The next point for consideration is 'What plants derive benefit from land plaster?' As a plant food one would suppose that it could only furnish directly those elements which it contains, viz., sulphur and lime. Indirectly, of course, it absorbs ammonia, which is a stimulant. Chemistry tells us that red clover contains a very large proportion of sulphuric acid and lime, and therefore we find by practical experience that red clover is the most beneficially affected of all plants. White clover contains an excess of potash, and again actual experience proves sulphate of potash produces the best effect upon that plant. But what chemical action takes place which produces this result on red clover? Here we reach into the domain of science and tread upon dangerous ground. However, the explanation given is that carbonate of ammonia and sulphate of lime cannot be brought together at common temperature without mutual decomposition. The ammonia enters into combination with the sulphuric acid and the carbonic acid with the lime, forming compounds that are not volatile, and which are retained in a condition serviceable to manure. It does not supply any actual deficiency, but acting as a sulphate or vegetable mold makes that available which already exists. It seems then that gypsum has two-fold functions. It acts, as stated, as a nutrient to certain plants, and directly by fixing the ammonia contained in the atmosphere, and in rain, dew and snow it furnishes a stimulant. We advise those who use plaster in clover culture to sow 100 lbs. per acre as soon as the ground will carry, and a second 100 lbs. when the plant is well headed out. We know that farmers are averse to this double labor, and particularly many of our young latter day farmers, who have no love for the operation of plaster sowing. But, in any case, plaster could be sowed at a sufficiently early date to catch the spring rains. Disappointment too often results from delay, especially when dry weather comes on and the plaster is not dissolved.

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